Chemistry Topics

INTRODUCTION TO CHEMISTRY

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Chemistry Is . . . ?

Matter and Energy

Measurement and the Metric System

Metric Prefixes

Scientific Notation

Volume and Density

Reporting Measured Quantities

Solving Problems

ATOMS, MOLECULES, AND IONS

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Introduction to the Atomic Model of Matter

Development of the Early Models of the Atom

The Current View of Atomic Structure

Identifying Elements: Names, Symbols, and Atomic

Numbers

Neutrons, Isotopes, and Mass Numbers

Molecules

Ions

THE ELECTRONIC STRUCTURE OF ATOMS

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Introduction

The Bohr Model of the Atom

The Modern (Wave-Mechanical) Model

Valence Electrons

Lewis Structures (Electron-Dot Diagrams)

Section II—Additional Material

Atomic Orbitals and Sublevels

Electron Configurations of Atoms

Lewis Structures and Atomic Orbitals

NUCLEAR CHEMISTRY

Section I-Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Nuclear Particles

Nuclear Equations

Natural Radioactivity and Radioactive Decay

Half-Life

Uses of Radioisotopes

Induced Nuclear Reactions

Section II-Additional Material

The Uranium-238 Decay Series

Isomeric Transition

Detection and Measurement of Radioactivity

Solving Radioactive Decay Problems

Particle Accelerators

Fission Reactors

THE PHASES OF MATTER

Section I Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Introduction

Gases

The Gas Laws

The Kinetic-Molecular Theory (KMT) of Gas

Behavior

Liquids

Solids

Change of Phase

Section II—Additional Material

Measuring Gas Pressure in the Laboratory

The Ideal (Universal) Gas Law

The Density of an Ideal Gas at STP

Gases and Chemical Reactions

Dalton's Law of Partial Pressures

Graham's Law of Effusion (Diffusion)

Gases Collected over Water

Additional Fusion and Vaporization Problems

Phase Diagrams

CHEMICAL PERIODICITY

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Introduction

The Periodic Table in History

The Modern Periodic Table

Properties Associated with Periodicity

Variation of Periodic Properties Among the

Elements

The Chemistry of the Representative Groups

The Chemistry of a Period

Section II—Additional Material

Sublevels and the Periodic Table

Successive Ionization Energies

Electron Affinity

Additional Aspects of First Ionization Energy

Variation of Successive Ionization Energies

Synthetic Elements

CHEMICAL BONDING AND MOLECULAR SHAPE

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Bonding and Stability

Ionic Bonding

Covalent Bonding

Electronegativity and Bonding

Drawing the Lewis Structures of Covalent Molecules

and Polyatomic Ions

Network Solids

Metallic Substances

Dipoles and Polar Molecules

Polarity and Molecular Symmetry

Intermolecular Forces

Physical and Chemical Properties of Bonded Substances:

A Summary

Section II-Additional Material

Resonance Structures

Additional Topics in Bonding

FORMULAS, EQUATIONS, AND CHEMICAL REACTIONS

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Chemical Formulas

Writing and Naming Chemical Formulas

Chemical Equations

Balancing a Chemical Equation

Classifying Chemical Reactions

Section II—Additional Material

Other Ways of Naming Ionic Compounds

SOLUTIONS AND THEIR PROPERTIES

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

A Solution Is . . . ?

Saturated and Unsaturated Solutions

Solubility

Concentrations of Solutions

Effect of the Solute on the Solvent

Behavior of Electrolytes in Solution

Section II—Additional Material

Mole Fraction

Molality

Dilution of Stock Solutions

Solutions and Chemical Equations

Calculating the Freezing and Boiling Points of

Solutions

Suspensions and Colloidal Dispersions

ENERGY AND CHEMICAL REACTIONS

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Energy and Its Measurement

Heat of Reaction

Potential Energy Diagrams

Spontaneous Reactions

Section II—Additional Material

Additional Calorimetry Problems

Transfer of Energy and Equilibrium Temperature

The Role of Energy in Chemical Reactions

Additional Aspects of Heats of Reaction

The Second Law of Thermodynamics

KINETICS AND EQUILIBRIUM

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Chemical Kinetics

Reversible Reactions and Dynamic Equilibrium

Phase Equilibrium

Solution Equilibrium .

Chemical Equilibrium

Section II-Additional Material

The Common-Ion Effect

Heterogeneous Equilibrium

The Equilibrium Constant (K_{en})

Problems Involving the Equilibrium Constant

Applications of Chemical Equilibrium

REDUCTION-OXIDATION (REDOX) AND ELECTROCHEMISTRY

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

What Are Oxidation and Reduction?

Formal Definitions of Oxidation and Reduction

Redox Equations

Spontaneous Redox Reactions

Electrochemical Cells

Section II—Additional Material

Balancing Redox Equations by the Half-Reaction

Method

Balancing Redox Equations by the Ion-Electron

Method

Half-Cell Potentials and Cell Voltage

The Standard Hydrogen Half-Cell

Electrolysis of Water and Aqueous NaCl (Brine)

Electroplating

Additional Applications of Redox and

Electrochemistry

ACIDS AND BASES

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Operational Definitions of Acids and Bases

Arrhenius Definitions of Acids and Bases

Acid-Base Titration

Brönsted-Lowry Definitions of Acids and Bases

The pH Scale of Acidity and Basicity

Acid-Base Indicators

Section II-Additional Material

Amphiprotic (Amphoteric) Substances

Acid-Base Equilibria

Conjugate Acid-Base Pairs

Neutralization (Revisited)

Strengths of Conjugate Acid-Base Pairs

Ionization Constants of Acids and Bases (K_a and K_b)

Ionization Constant of Water (K_{yy})

A More Detailed Look at pH and pOH

Hydrolysis of Salts in Aqueous Solutions

Acid-Base Properties of Oxides

Lewis Definitions of Acids and Bases

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ORGANIC CHEMISTRY

Section I—Basic (Regents-Level) Material

NYS Regents Concepts and Skills

Organic Chemistry Is . . . ?

Comparison of Organic and Inorganic Compounds

Hydrocarbons and Homologous Series

Functional Groups

Organic Reactions

Section II—Additional Material

Stereoisomerism

The Benzene Series

Primary, Secondary, and Tertiary Alcohols

Dihydroxy and Trihodroxy Alcohols

Types of Polymerization